

Pharmacophore-Based Design of Sphingosine 1-phosphate-3 Receptor Antagonists That Include a 3,4-Dialkoxybenzophenone Scaffold

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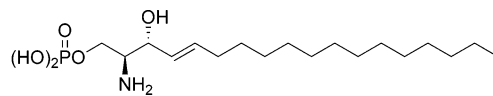
Sphingosine 1-phosphate (S1P) receptors are G-protein-coupled receptors. Among the five identified subtypes S1P₁–5, the S1P₃ receptor expressed on vascular endothelial cells has been shown to play an important role in cell proliferation, migration, and inflammation. A pharmacophore-based database search was used to identify a potent scaffold for an S1P₃ receptor antagonist by common feature-based alignment and further validated using the Güner–Henry (GH) scoring method. Assumed excluded volumes were merged into this model to evaluate the steric effect with the S1P₃ receptor. Three commercially available compounds were identified as S1P₃ receptor antagonists, with IC₅₀ values <5 μM. The synthesis of further derivatives revealed that the 3,4-dialkoxybenzophenone scaffold is a potent component of an S1P₃ receptor antagonist. Our results indicate that pharmacophore-based design of S1P₃ receptor antagonists can be used to expand the possibility of structural modification through scaffold-hopping based on a database search.

Introduction

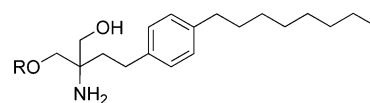
Sphingosine 1-phosphate (S1P⁴) is a potent lipid mediator produced from the metabolism of sphingomyelin (Figure 1). S1P acts on a family of G-protein-coupled receptors (GPCRs) and transduces intracellular signals involved in numerous cellular processes.^{1,2} The S1P₁ receptor, which was the first S1P receptor to be identified, was initially named endothelium differentiation gene-1 (Edg-1) receptor.³ The five subtypes of S1P-induced Edg receptors were renamed S1P receptors, consistent with the official IUPHAR nomenclature for rational classification; S1P₁ (formerly Edg-1), S1P₂ (formerly Edg-5), S1P₃ (formerly Edg-3), S1P₄ (formerly Edg-6), and S1P₅ (formerly Edg-8).⁴

S1P₃ receptors expressed on vascular endothelial cells play several important roles, such as in cell proliferation based on extracellular signal-regulated kinase (ERK) activation in a pertussis toxin (PTX)-sensitive manner,⁵ developmental and pathological angiogenesis,⁶ endothelial cell migration,⁷ and the activation of eNOS.⁸ The role of vascular S1P₃ receptors has been reviewed in detail.^{9–12}

The advent of the novel immunosuppressant **2** has inspired considerable interest in agonists and antagonists of S1P receptors (Figure 1).¹³ It has been demonstrated that **2** is metabolized across various species to a monophosphate ester **3**, which is a high-affinity ligand for S1P₁, S1P₃, S1P₄, and S1P₅, but not S1P₂.^{14,15} Administration of **2** resulted in the sequestration of circulating lymphocytes within secondary lymphoid tissues. Its egress-blocking effect occurs through the inactivation of the S1P₁ receptor.¹⁶ Therefore, an S1P₁ receptor-selective agonist or antagonist has been expected to be a more effective immunosuppressant than **2**.

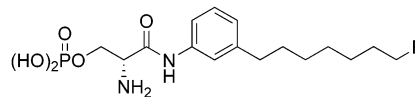


Sphingosine 1-phosphate (S1P) **1**



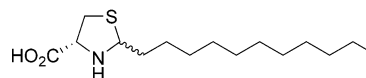
R = H **2** (FTY720)

R = -P(=O)(OH)₂ **3** (FTY720 phosphate (FTY720-P))



R = Me **4** (VPC23019)

R = H **5** (VPC25239)



(4R)-4-Undecylthiazolidine-2-carboxylic acid **6**

Figure 1. S1P, reported S1P₁ receptor agonists and S1P₃ receptor antagonists (**3**, S1P_{1, 3–5} receptor agonist; **2**, immunosuppressant which is the precursor for **3**; **4** and **5**, S1P₁/S1P₃ receptor antagonists; **6**, S1P₃ receptor antagonist).

Clinical studies with **2** have identified dose-dependent transient symptomatic bradycardia in stable renal transplant patients.¹⁷ Bradycardia is consistent with the S1P-induced activation of muscarinic receptor-activated inwardly rectifying K⁺ current (*I*_{K,Ach}).^{18–20} Recent studies in S1P₃ null mice have shown that the induction of bradycardia is mediated through the activation of S1P₃ receptors.^{21,22} Meanwhile, S1P₂ and S1P₃, which are abundantly expressed in airway smooth muscle

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^a Abbreviations: S1P, sphingosine 1-phosphate; EDG, endothelium differentiation gene; GH score, Güner–Henry score.

(ASM), mobilize calcium and mediate airway hypercontractility.²³ Administration of **2** induces bronchoconstriction and elevates airway constriction, while administration of an S1P₁ receptor-selective agonist has no effect on bradycardia or airway resistance.^{24,25}

Taken together, these results suggest that an S1P₃ receptor antagonist may be useful for preventing vascular proliferation, bradycardia, and bronchoconstriction. Furthermore, elucidation of the structural requirements for S1P₃ receptor antagonist may enable the development of a more selective S1P₁ receptor antagonist. While several S1P₁ receptor-selective agonists have been reported, there have been few reports on S1P₃ antagonists.²⁶ Although suramin has been reported to be an S1P₃ receptor-selective antagonist, it also acts as a reversible P2-purinoreceptor antagonist,^{27,28} an inhibitor of DNA topoisomerase II,²⁹ and a potent inhibitor of the reverse transcriptase of RNA tumor viruses.³⁰ Suramin has also been reported to act as a direct antagonist of heterotrimeric G proteins.³¹ Recently, practical S1P₁/S1P₃ receptor antagonists **4** and **5**, which have meta-alkylated phenyl structures, were reported from the University of Virginia (Figure 1).^{32,33} However, the activities of these antagonists toward S1P₃ receptors are at least one-fifth less than those toward S1P₁ receptors.

To identify structurally novel agonists or antagonists efficiently, a computational approach for scaffold-hopping has been extensively applied.³⁴ Structure-based virtual screening, in which the 3D structure of a target protein (usually derived from X-ray crystallography) is used in protein–ligand docking calculations, has been used extensively to explore potent compounds. However, this approach to scaffold-hopping in GPCRs is limited due to issues involving purification and crystallization. Meanwhile, a pharmacophore is a generic description of the key features of an active molecule and is usually built by overlaying different conformations of active compounds. Pharmacophore models derived from active ligands have been widely used to explore GPCR ligands.³⁵

We developed 2-alkyl or aryl thiazolidine-4-carboxylic acid derivatives as potent S1P₃ antagonists at 10 μ M by a pharmacophore-based database search which included information on the conformations of S1P.³⁶ However, very recently, one of these compounds, (4*R*)-4-undecylthiazolidine-2-carboxylic acid **6** (Figure 1), has been reported to be inappropriate for identifying selective antagonism at the S1P₃ receptor due to its low potency and reduced selectivity.³⁷ More potent and practical S1P₃ receptor antagonists are still needed.

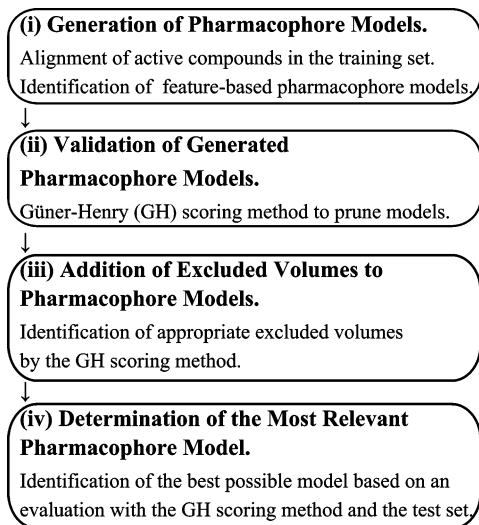
We have continued to study more potent S1P₃ receptor antagonists, which may be useful for influencing vascular proliferation and inflammation, or at least for elucidating the detailed mechanism of S1P₃ receptor antagonism. In this report, we describe (i) the generation and assessment of a pharmacophore model for S1P₃ receptor antagonists, (ii) the identification of novel potent S1P₃ antagonists retrieved from among commercially available compounds by a pharmacophore-based database search, (iii) the structure–activity relationship of synthesized derivatives as S1P₃ receptor antagonists, and (iv) the structural requirements for S1P₃ receptor antagonists.

Methods

1. Generation and Validation of Pharmacophore Models.

The scheme for generating reliable pharmacophore models for S1P₃ receptor antagonists is outlined in Chart 1. First, pharmacophore models were constructed by aligning active compounds as S1P₃ receptor antagonists with multiple conformations (training set). Second, the generated pharmacophore models

Chart 1. Chart for Deriving a Pharmacophore Model of S1P₃ Receptor Antagonists



were validated to identify preferable models that could be used to efficiently exclude inactive compounds. Third, information regarding steric properties was added to the pharmacophore models, since these models could not exclude inactive compounds which probably had decreased potency due to extreme molecular volumes. Finally, these modified pharmacophore models were validated if they made it possible to efficiently exclude inactive compounds and to identify active compounds that were not used for pharmacophore generation (test set). The most reliable pharmacophore model was then identified. The details of each step are described below.

(i) Generation of the Pharmacophore Model. A feature-based pharmacophore model derived from the active compound set with diverse conformations without quantitative activity data was applied.³⁸ It describes possible pharmacophoric space as an arrangement of features. For example, a feature, such as *hydrogen-bond acceptor*, is described which defines the kind of interaction with the target protein.

In this paper, we used three common features: *hydrogen-bond acceptor*, *hydrophobic*, and *negative ionizable*. The *hydrophobic* feature matches the following types of groups of atoms: a contiguous set of atoms that are not adjacent to any concentrations of charge (charged atoms or electronegative atoms) in a conformer such that the atoms have surface accessibility, including phenyl, cycloalkyl, isopropyl, and methyl groups. The *hydrogen-bond acceptor* feature matches the following types of atoms or groups of atoms with surface accessibility: sp or sp² nitrogens that have a lone pair and charge less than or equal to zero, sp³ oxygens or sulfurs that have a lone pair and charge less than or equal to zero, or nonbasic amines that have a lone pair. The *negative ionizable* feature matches atoms or groups of atoms that are likely to be deprotonated at physiological pH, such as trifluoromethyl sulfonamide hydrogens, sulfonic acids, phosphonic acids, sulfinic, carboxylic, or phosphinic acids, tetrazoles, or negative charges not adjacent to a positive charge.

(ii) Validation of the Quality of Generated Pharmacophore Models. The Güner–Henry (GH) scoring method^{39–41} was used to assess the quality of pharmacophore models. The GH score is a metric for quantifying the precision of hits and the recall of actives mined from a database consisting of known actives and inactives (Chart 2). It is considered a relevant metric, since it takes into account both the percent yield of actives in a

Chart 2. Metrics for Analyzing Hit Lists by a Pharmacophore Model-Based Database Search^a

$$\% A = \frac{H_a}{A} \times 100$$

$$\% Y = \frac{H_a}{H_t} \times 100$$

$$E = \frac{H_a / H_t}{A / D}$$

$$GH = \left(\frac{H_a (3A + H_t)}{4H_t A} \right) \left(1 - \frac{H_t - H_a}{D - A} \right)$$

^a H_a , number of actives in the hit list (true positives); H_t , number of hits retrieved; A , number of active compounds in the database; D , number of compounds in the database; $\% A$, ratio of actives retrieved in the hit list (precision); $\% Y$, yield of actives (recall); E , enrichment of the concentration of actives by the model relative to random screening without any pharmacophoric approach; GH, Güner–Henry score.

database ($\% Y$, recall) and the percent ratio of actives in the hit list ($\% A$, precision). The GH score ranges from 0, which indicates the null model, to 1, which indicates the ideal model. On the basis of S1P-induced cytosolic Ca^{2+} mobilization in CHO cells expressing the S1P₃ receptor, a compound with more than 30% inhibitory activity at 10 μM was defined as an “active” and a compound with less than 20% inhibitory activity at 10 μM was defined as an “inactive”. For GH score assessment, all compounds with inhibitory activities of 20 to 30% at 10 μM were excluded to simplify the classification.

(iii) Addition of Excluded Volumes to Pharmacophore Models. Steric interaction can be approximated by the strategic placement of excluded volumes (spheres that cannot be penetrated by the ligand) in the pharmacophore model. The inclusion of excluded volumes in pharmacophore models has been reported to improve the pruning of database hit lists by 30–75%.⁴² To obtain a more desirable pharmacophore model, appropriate excluded volumes were added by the GH scoring method.

(iv) Determination of the Most Relevant Pharmacophore Model. Pharmacophore models with excluded volumes were validated to identify the most relevant model by the same GH scoring method as described in section ii. In addition, pharmacophore models that could not identify any compound in the test set were excluded.

2. Database Searches. Composite databases, which contained a total of approximately three million commercially available compounds, were searched using the pharmacophore model. Hit compounds (hit list) can be ranked according to the fit value, which is the degree of consistency with the pharmacophore model. In this paper, the fit value is defined as the relative percentage compared to the maximum fit value. To decrease the number of hits, a minimum fit value, which is the lowest limit to qualify as a hit compound, was applied. In this paper, we used 90% of the minimum fit value to prune the hit list.

3. Biological Assay. The IC₅₀ values for S1P₁, S1P₂, and S1P₃ receptors were quantitatively determined in terms of the Fura-2 fluorescent intensity. Dye was loaded into CHO cells that stably expressed S1P₁, S1P₂ or S1P₃ receptors. After addition of an antagonist, the inhibition of the transient increase in the intracellular calcium ion concentration ($[Ca^{2+}]_i$) induced by S1P was measured.

Results and Discussion

1. Generation and Validation of Pharmacophore Models.

(i) Generation of Pharmacophore Models. Pharmacophore

models for S1P₃ receptor antagonists were generated using five of seven active compounds: **6**, **8**, **10**, **11** and **12**, which were previously identified (Figure 2).³⁶ Compounds **7** and **9** were used as the test set in validation. Twelve patterns of pharmacophore generation were applied, based on three major parameters (Table 1): **(i)** The number of maximum features applied in a pharmacophore model was varied from three to five. **(ii)** *Negative ionizable* and *hydrophobic* features were applied for the generation of all models. Meanwhile, models were separately generated with or without the *hydrogen-bond acceptor* feature, since it was not clear whether this feature was important. **(iii)** 2-Alkyl or aryl thiazolidine-4-carboxylic acid derivatives **6**, **8**, **10** and **11** in the training set are diastereomers at C2 ((2*R*)- and (2*S*)-isomer). This complication may hamper conformational analysis and subsequent pharmacophore generation. Therefore, two training sets that included (2*R*)-thiazolidine-4-carboxylic acid derivatives **6**, **8**, **10** and **11** or their (2*S*)-derivatives were used. Compound **12** was used in both training sets. As a result, these three conditions gave 307 pharmacophore models.

(ii) Validation of Generated Pharmacophore Models. The quality of pharmacophore models was assessed using the GH score as a metric to search a database that consisted of 100 inactive compounds and seven active compounds **6–12** (all of their structures are shown in the Supporting Information). Since the results of validation were strongly influenced by the quality of the database, the composition of the active and inactive compounds in the database is the predominant factor in GH score assessment. Thus, the inclusion of inactive compounds was limited according to the following criteria. **(A)** All inactive compounds should match at least three types of features (*negative ionizable*, *hydrophobic* and *hydrogen-bond acceptor* features), the same as active compounds, since an inactive compound without one of these features in a pharmacophore model is excluded easily and thus it cannot reflect its ability. **(B)** Eighteen inactive compounds with thiazolidine-4-carboxylic acid scaffolds were included to avoid obtaining a too-strict and rigid pharmacophore model that could identify only the thiazolidine scaffold as an S1P₃ receptor antagonist.

The following criteria were applied to identify desirable pharmacophore models: A GH score of more than 0.4 was used to limit the number of models by less than 50%. A recall value of more than 0.5 was used for desirable pharmacophore models. Three hundred seven pharmacophore models were analyzed in a scatter diagram between the GH score and the recall of active compounds (Figure 3). Plots were divided into four blocks by the above criteria. Most of the plots were in the lower squares, which indicated a GH score of less than 0.4 and that these were nonselective pharmacophore models. The upper-left square showed too-exclusive pharmacophore models that could only identify a few active compounds. Meanwhile, in the upper-right square (GH score of more than 0.4 and recall of 0.5), the pharmacophore models can selectively retrieve most active compounds. The seven pharmacophore models in this upper-right square, which fulfilled the criteria, were selected from among the 307 models (Figure 4).

Pharmacophore models R1, R2, R3, and R4 were generated by the training set that included (2*R*)-thiazolidine-4-carboxylic acid derivatives. Meanwhile, pharmacophore models S1, S2 and S3 were generated by the model that consisted of (2*S*)-derivatives. All seven models included four features: one *negative ionizable*, one *hydrogen-bond acceptor*, and two *hydrophobic* features.

(iii) Addition of Excluded Volumes to Pharmacophore Models by Considering the GH Score. Our previous study

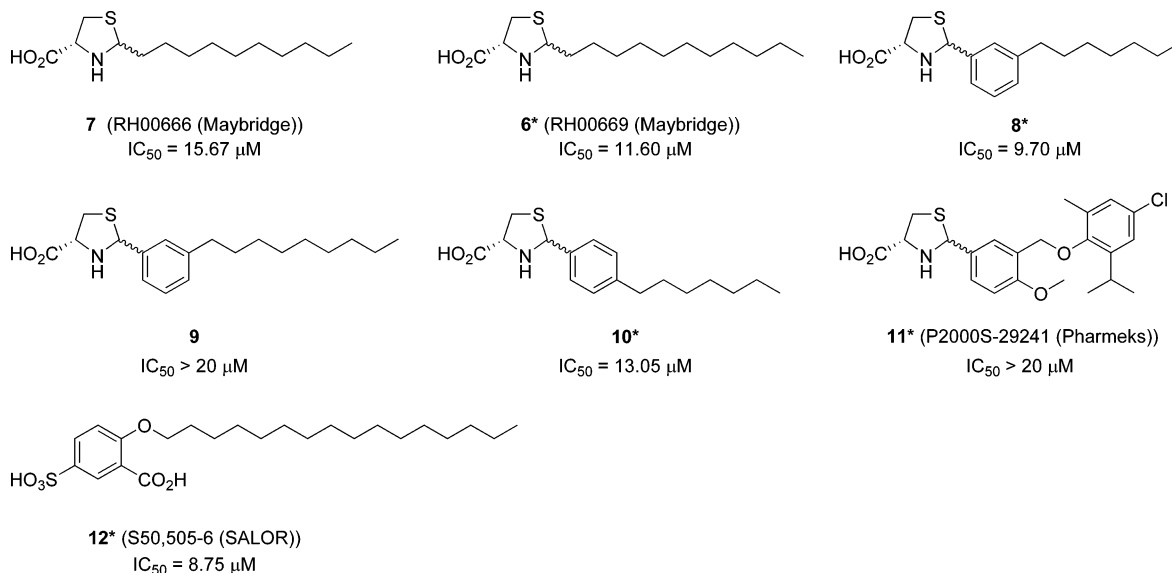


Figure 2. Previously identified S1P₃ receptor antagonists. Commercially available compounds are shown by the catalog number with the supplier's name in parentheses. The IC_{50} value was measured by the inhibition of S1P-induced cytosolic Ca²⁺ mobilization in CHO cells expressing S1P₃ receptor in terms of Fura-2 fluorescence (see Experimental Section). For compounds **9** and **11**, while IC_{50} values were over 20 μM , dose-dependent inhibitions were observed. The asterisk (*) represents the training set for S1P₃ receptor antagonists that was used to generate pharmacophore models.

Table 1. Conditions for Pharmacophore Generation along with the Number of Features, with or without the Hydrogen-Bond Acceptor Feature and the Chirality at C2 of the (2*R*)- or (2*S*)-Thiazolidine Scaffold in the Training Set

NegI	number of features ^a		number of models ^b	
	HBA	HYD	2 <i>R</i>	2 <i>S</i>
1	1	1	18	17
1	0	2	15	12
1	1	2	60	49
1	0	3	17	15
1	1	3	52	47
1	0	4	0	5

^a Number of maximum features used in pharmacophore generation. NegI, negative ionizable feature; HBA, hydrogen-bond acceptor feature; HYD, hydrophobic feature. ^b Number of models generated by using a training set that included (2*R*)- or (2*S*)-thiazolidine-4-carboxylic acid derivatives.

showed that the S1P₃ antagonist activity of thiazolidine-4-carboxylic acid was increased by lengthening the side alkyl chain at the 2-position up to 11 methylenic groups and then decreased as the number of methylenic groups increased further.³⁶ One possible explanation for this observation is based on the steric effects of the methyl group C13 in 2-tridecyl thiazolidine-4-carboxylic acid **13** (Figure 5). All seven models in Figure 4 identified **13** as an active compound due to the lack of information about steric properties. Therefore, these models should be improved to take into account steric properties and then further validated.

Overlap between the pharmacophore model and **13** could reveal the 3D-coordinates of C13 positions which are expected as excluded volumes. For that purpose, the following methods were used: (A) All of the generated conformations of **13** with the (2*R*)- or (2*S*)-isomer were superimposed on each pharmacophore model derived from the training set consisting of (2*R*)- or (2*S*)-derivatives, respectively. All of the 3D-coordinates of C13 of **13** were transferred to the center coordinates of excluded volumes (spheres) with radius 1.5 Å and merged with the pharmacophore model. (B) All active compounds **6–12** were aligned with each pharmacophore model with the excluded volumes. If an excluded volume sphere partially overlapped any sphere of an atom in active compounds with radius 1.5 Å, this

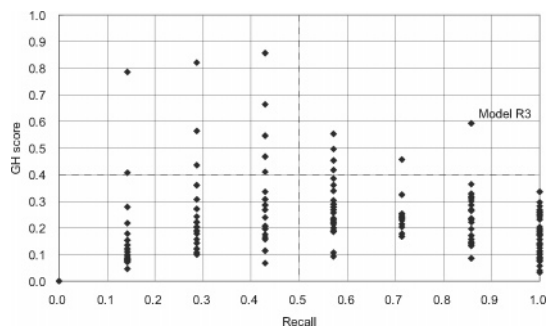


Figure 3. Scatter-diagram analysis of pharmacophore model validations based on the GH score and recall of active compounds. A GH score of 1 means the best (ideal) model and 0 means the worst (null) model. Recall means the yield of known active compounds registered in a database (also called yield of actives (% *Y*)). For example, if a pharmacophore model identifies all seven known actives **6–12** as hits among 107 compounds, the recall value is 1 (100%).

excluded volume was discarded. Consequently, excluded volumes outside the active space were retained. For example, in model R3, 95 excluded volumes surrounding the active space were retained (Figure 6a). (C) Each excluded volume was assessed by the GH score using the above-mentioned database. Only excluded volumes that improved the GH score were retained, since they could filter out inactive compounds, probably based on steric effects. For example, in model R3, three of the 95 excluded volumes were assigned to be associated with steric interaction based on improvement of the GH score (Figure 6b,c).

(iv) Determination of the Most Relevant Pharmacophore Model. As an S1P₃ antagonist pharmacophore model, model R3 with three excluded volumes (R3 Ex) was chosen among seven models with excluded volumes based on the GH score (Table 2).

The GH score of model R3 without an excluded volume was 0.59, which was the best among the seven models, and with three excluded volumes this value increased to 0.93, which was also the best among the seven models with excluded volumes. In addition, model R3 could identify both **7** and **9** in the test set that were not used for pharmacophore generation.

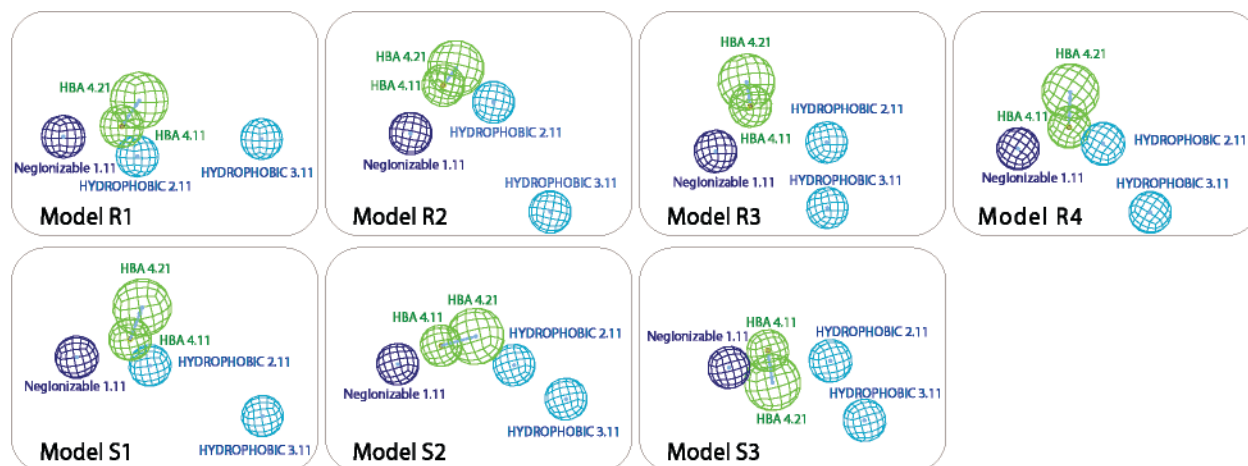


Figure 4. Seven pharmacophore models that consisted of one *negative ionizable*, one *hydrogen-bond acceptor*, and two *hydrophobic* features fulfilled the criteria. The letters R and S with each name indicate the pharmacophore model generated by a training set consisting of (2*R*)- and (2*S*)-thiazolidine derivatives, respectively. Four features: NegIonizable (*negative ionizable*), HYDROPHOBIC (*hydrophobic*) 2.11 and 3.11, and HBA (*hydrogen-bond acceptor*) 4.11 features are represented as spheres with radius 1.72 Å. The HBA (*hydrogen-bond acceptor*) 4.21 feature is the projected point of HBA 4.11 and is represented as a sphere with radius 2.32 Å.

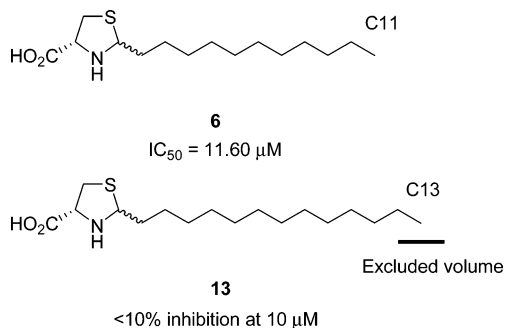


Figure 5. Assumed steric repulsion between SIP_3 receptor and **13**. As a possible explanation for the significant decrease in the potency of **13** compared to **6**, steric repulsion with the SIP_3 receptor can be approximated at C13 of **13**. In a pharmacophore model, a kind of steric repulsion can be represented by the strategic placement of excluded volume (a sphere that cannot be penetrated by the ligand).

2. Database Searches To Identify SIP_3 Receptor Antagonists, Biological Assay, and Results. Model R3 Ex was used to search databases consisting of commercially available compounds. Among the hit list, 36 compounds were selected, purchased, and assayed to identify SIP_3 antagonist activity (the structures of all assayed compounds are shown in the Supporting Information).

Among the screened compounds, isophthalic acid derivatives **14**, **15**, and **16** were identified as antagonists for SIP_3 receptor (Figure 7). While they showed similar potencies, we chose isophthalic acid derivative **14** for further studies to clarify the structural requirements because it contained fewer carboxylic acids than **15** and **16**. Isophthalic acid **14** was aligned to model R3 Ex (Figure 8a). One of the carboxylic acids on the phenyl ring was recognized as a *negative ionizable* feature. The other carboxylic acid was assumed to act as the *hydrogen-bond acceptor* feature. The 3-methoxyphenyl group, but not the 4-cetyloxy group, was mapped as a *hydrophobic* feature. The direction of the 4-cetyloxy group was different from that of the *m*-substituted alkyl chain in **8** (Figure 8a,b). The piperidine group, which was recognized as a *hydrophobic* feature, was also considered to play an important role in regulating the active conformation.

3. Synthesis of Derivatives and Identification of the Structure–Activity Relationship in an SIP_3 Receptor Antagonist. To clarify the minimum structural requirements for

14 as an SIP_3 receptor antagonist, derivatives of **14** were synthesized (Scheme 1). We replaced the piperidine group with a phenyl group to simplify synthesis, since we assumed that it acted as a *hydrophobic* feature.

Compound **19** or **20** was coupled with the benzophenone derivative to elucidate whether two carboxylic acids were essential, as the model predicted. Five benzophenone building blocks **21**–**24** were synthesized to clarify the substituent effects of 4-cetyloxy and 3-methoxy groups on the phenyl ring. Both geometric isomers (*E* and *Z* isomers) with regard to the hydrazone moiety were obtained by a coupling reaction. They were purified separately as single isomers by column chromatography. The geometry of the hydrazone bond was assigned by comparison of the ^{13}C NMR chemical shift of the hydrazone carbon of the *E* and *Z* isomers. A steric compression shift that results in an upfield shift is observed in sterically hindered *Z* isomer of oxime or hydrazone by ^{13}C NMR spectroscopy^{43–46} and has been used in the studies of diverse organic compounds.^{47–50} For example, the ^{13}C NMR chemical shift of the hydrazone carbon of *E* isomer **25E** was 150.04 ppm and that of *Z* isomer **25Z** was 149.26 ppm. Thus, a steric compression shift of 0.78 ppm was observed for the *Z* isomer. Compounds **25**–**29** were purified separately as *E* or *Z* isomers. We obtained the corresponding carboxylic acids **30**–**34** by hydrolysis.

SIP_3 receptor antagonist activities of these synthesized derivatives **30**–**34** were evaluated (Table 3). Isophthalic acid **14** retrieved from the database showed an IC_{50} value of 3.44 μM , and this also showed moderate antagonist activity for the SIP_1 receptor. Synthesized **30E** showed an IC_{50} value of 2.96 μM . We also synthesized a derivative without this phenyl group, corresponding to compound **30E**; however, SIP_3 antagonist activity was not observed at all (data not shown). Therefore, this phenyl group was implicated in the interaction with the SIP_3 receptor. None of the synthesized sulfonylhydrazone derivatives showed more than 50% inhibitory activity for the SIP_2 receptor at 10 μM .

In the assay of **32E**, **31E**, and **30E** with 4, 8, and 16 alkyl chain carbons, respectively, the IC_{50} values for the SIP_3 receptor and the SIP_1 receptor were found to decrease with a decrease in the length of the alkyl chain. Therefore, the length of the alkyl chain at the 4-position of the phenyl ring was the dominant factor for not only the SIP_3 receptor but also SIP_1 receptor

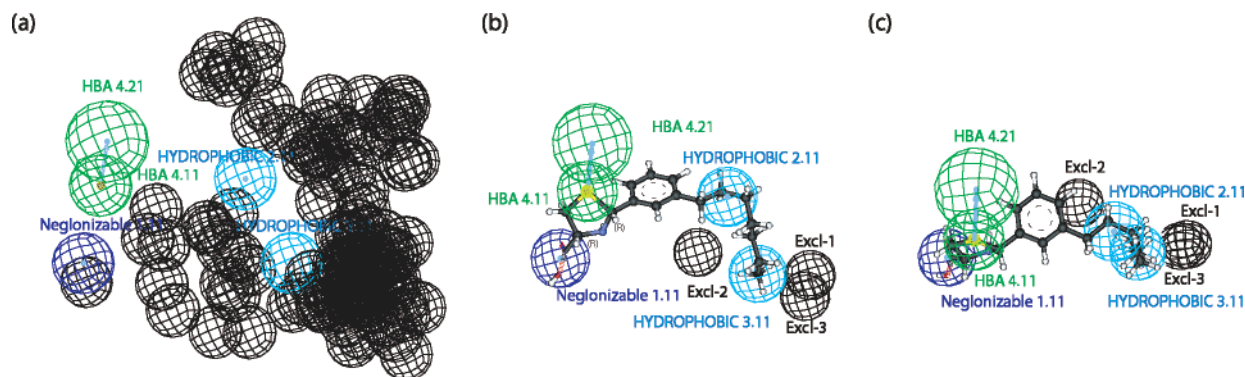


Figure 6. Addition of all excluded volumes to model R3 to identify **13** as inactive. Overlap between the optimized model (R3 Ex) and **8**. (a) Ninety-five 3D coordinates of C13 in compound **13** aligned with model R3 were transferred to the corresponding excluded volumes with radius 1.5 Å and merged into that model. (b and c) Viewpoints of model R3 Ex aligned with compound **8**. Three excluded volumes 1–3 are abbreviated as Excl-1, -2, and -3 and indicated with black mesh.

Table 2. Comparison of Database Searches Using Models R3 and R3 Ex^a

model	all	actives	hits	tp	fp	tn	fn	%A	%Y	E	GH
R3	107	7	11	6	5	95	1	54.5	85.7	8.34	0.59
R3 Ex	107	7	5	5	0	100	2	100	71.4	15.3	0.93

^a All, number of compounds in the database; actives, number of actives in the database; hits, number of hits retrieved by the model (minimum fit value of 90%); tp, number of true positives in the hit list; fp, number of false positives in the hit list; tn, number of true negatives in the hit list; fn, number of false negatives in the hit list.

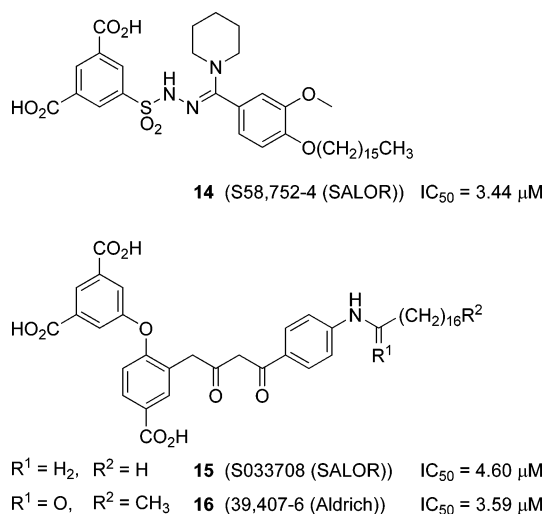


Figure 7. Newly identified S1P₃ receptor antagonists obtained via a database search. The supplier's name is given in parentheses. The IC_{50} value was measured by the inhibition of S1P-induced $[Ca^{2+}]_i$ mobilization in CHO cells expressing the S1P₃ receptor in terms of Fura-2 fluorescence.

antagonist activity. This tendency was also observed for **31Z** and **30Z**.

As expected based on the pharmacophore model, the carboxylic acid of isophthalic acid, which was identified as a *hydrogen-bond acceptor* feature, was also essential for S1P₃ antagonist activity, since benzoic acid derivatives **34E** and **34Z** did not show clear potencies. Compound **30Z** exhibited a similar potency (IC_{50} 3.57 μM) to **30E**. Both of the active conformations **30E** and **30Z** were compared along with model R3 Ex (Figure 9a,b). 3,4-Dialkoxybenzophenone acted as *hydrophobic* 2.11 and 3.11 in both **30E** and **30Z**. The 3-methoxy group of **30E** was superimposed on *hydrophobic* 3.11, as in compound **14**. Meanwhile, the 3-methoxy group of **30Z** could also be superimposed on *hydrophobic* 3.11, since the symmetry of isophthalic acid allowed each carboxylic acid to act as either a *negative ionizable* or *hydrogen-bond acceptor* feature, with the result that **30Z** could be superimposed on the model similar to

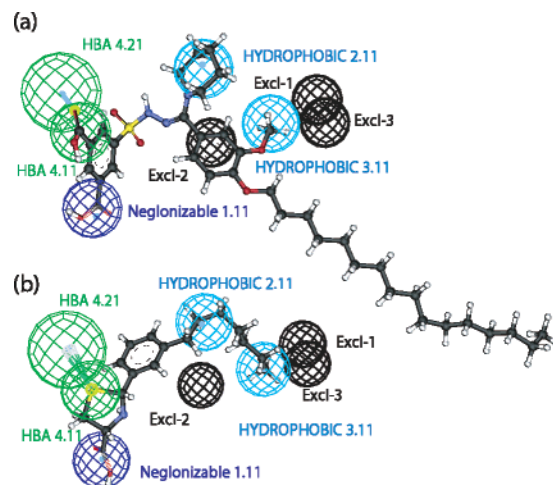
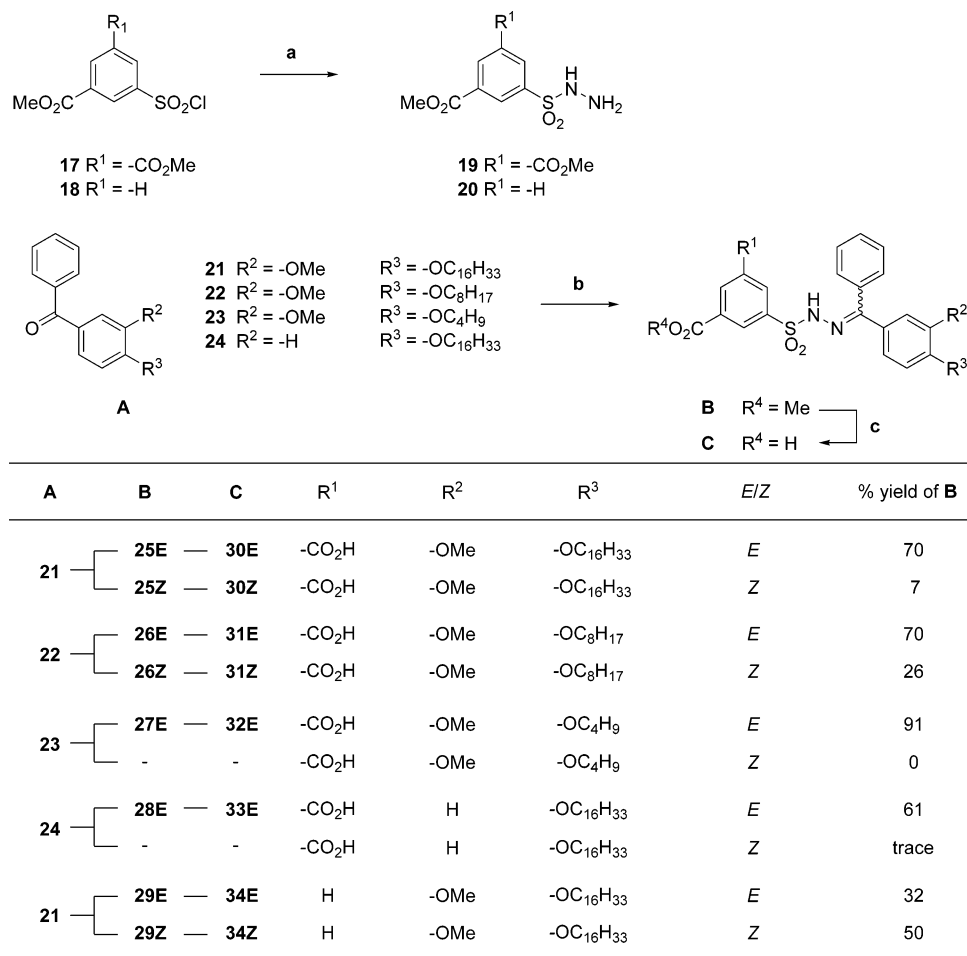


Figure 8. Overlap between model R3 Ex and **14** and **8**. (a) **14** with model R3 Ex. (b) **8** with model R3 Ex.

30E. This assumption regarding the pharmacophore model is consistent with the results of the biological assay. The importance of the 3-methoxy group as the *hydrophobic* feature was supported by the finding that compound **33E**, which does not contain a 3-methoxy group, was less potent than compound **30E**.

As mentioned earlier, the 4-cetyloxy group did not correspond to the meta-substituted alkyl chain of compound **8** (Figure 8a,b). However, the 4-cetyloxy group in isophthalic acid derivatives was significantly associated with antagonist activities for S1P₁ and S1P₃ receptors. Taken together, these results suggest that the 4-cetyloxy group plays a role corresponding to the alkyl chain of S1P. Initially, we hypothesized that the hydrophobic space represented by *hydrophobic* 2.11 and 3.11 corresponded to the alkyl chain of S1P. However, we now believe that it works differently than S1P and is significantly associated with antagonist activity for the S1P₃ receptor. Thus, a 3,4-dialkoxybenzophenone scaffold which fulfilled this hydrophobic space could be a potent component of an S1P₃ receptor antagonist. Meanwhile, since a 4-cetyloxy group may play a role similar to that of the alkyl chain of S1P, it is assumed to contribute to

Scheme 1. Synthesis of 5-sulfohyrazono-1,3-isophthalic Acid Derivatives and 3-Sulfohyrazonobenzoic Acid Derivatives^a

^a Reagents and conditions: (a) NH₂NH₂·H₂O, THF, 0 °C; (b) **19** or **20**, MeOH, reflux; (c) 1N NaOH, THF–MeOH.

Table 3. S1P₃ Receptor Antagonist Activities of Synthesized Isophthalic Acid Derivatives^a

compd	IC ₅₀ (μM)		
	S1P ₃	S1P ₁	S1P ₂
14	3.44	6.18	(35.9%)
30E	2.96	6.64	(40.9%)
30Z	3.57	4.28	(9.5%)
31E	11.39	(12.7%)	(-4.6%)
31Z	13.23	(32.9%)	(14.0%)
32E	(6.7%)	(24.2%)	(6.2%)
33E	9.74	(17.2%)	(20.5%)
34E	(34.4%)	(34.8%)	(-0.1%)
34Z	(8.3%)	(1.0%)	(-8.7%)

^a The IC₅₀ value was measured by the inhibition of S1P-induced [Ca²⁺]_i mobilization in CHO cells expressing S1P₁, S1P₂, or S1P₃ receptor in terms of Fura-2 fluorescence. For a compound with lower potency, for which it was difficult to measure an accurate IC₅₀ value, the percent inhibition at 10 μM is given in parentheses.

all S1P receptors and to lack selectivity for S1P₃ receptor antagonism. Thus, **31E** with a moderate alkyl chain may be useful not only for exploring a more selective pharmacophore model for S1P₃ receptor antagonists but also as a better starting point to obtain a more potent and selective S1P₃ receptor antagonist than **14** and **30E**.

4. Assessment of the Model Using Different S1P₃ Receptor Antagonists. Although the pharmacophore model identified a potent scaffold, it was based on a training set that consisted of compounds with limited diversity and activities. To generate a feature-based pharmacophore model, structural diversity is

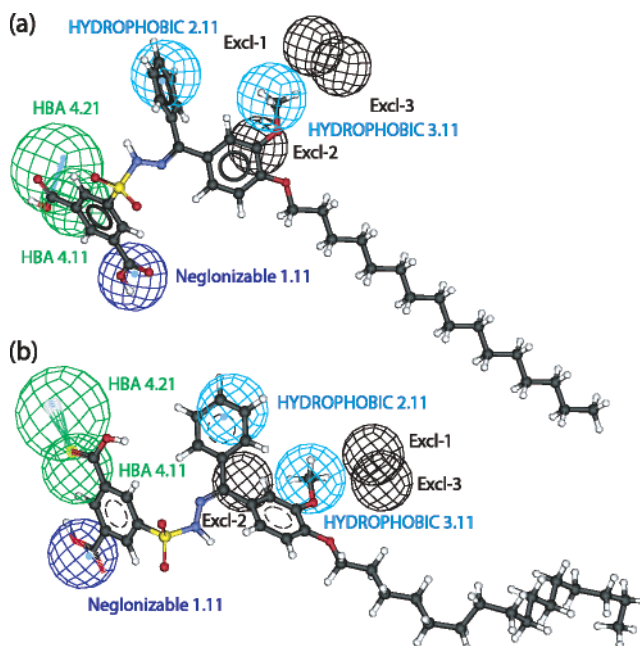


Figure 9. Overlap analysis of *E* and *Z* isomers. (a) **30E** (*E* isomer) with model R3 Ex. (b) **30Z** (*Z* isomer) with model R3 Ex.

important for exploring the key structural features. In addition, there were too few compounds in the test set. Thus, the pharmacophore model was further evaluated using different S1P₃

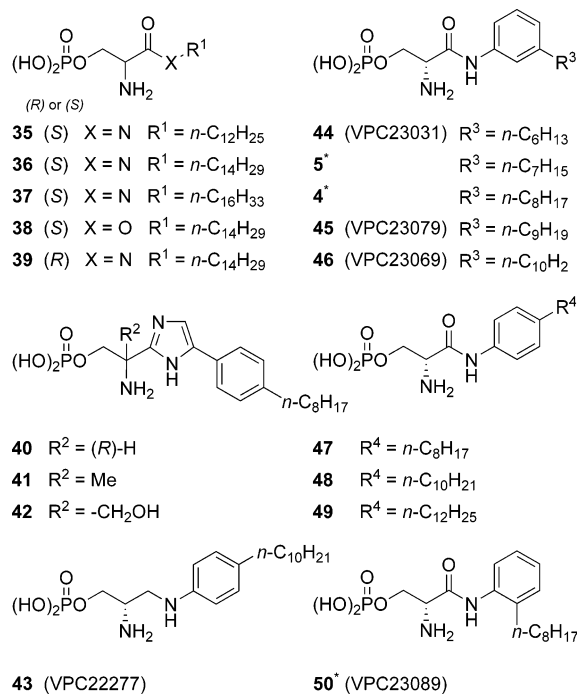


Figure 10. SIP₁/SIP₃ receptor agonists and antagonists reported by the University of Virginia. The three asterisks (*) represent the reported SIP₁/SIP₃ receptor antagonists, and their antagonist activities for the SIP₃ receptor were determined by a [³²P]-SIP binding assay (**4**; K_i = 1175 nM, **5**; K_i = 98 nM, **50**; K_i = 1586 nM).³² Compounds **35–42** and **47–49** were reported to be SIP₁/SIP₃ receptor agonists with EC₅₀ values for the SIP₃ receptor that were greater than 1000 nM based on a [³⁵S]-γ-GTP binding assay.^{51,52} Compounds **43** and **46** were reported to be both SIP₁ receptor agonists and SIP₃ receptor partial agonist and weak partial agonist, respectively.³² Compound **44** was reported to be an SIP₁ receptor antagonist and **45** was reported as an SIP₁/SIP₄ receptor agonist.³²

receptor antagonists to see whether the model could identify the same key features for SIP₃ receptor antagonism in our result.

Recently, researchers at the University of Virginia reported that phosphates **4**, **5**, and **50** were SIP₁/SIP₃ receptor antagonists based on the result of [³⁵S]-γ-GTP binding and [³²P]-SIP binding assays (Figure 10).³² Meanwhile, most of the derivatives in Figure 10 were reported to act as SIP₃ receptor agonists.^{32,51,52} They were useful for assessing the potency of the pharmacophore model for SIP₃ receptor antagonists, since they included agonists (**35–43** and **46–49**), antagonists (**4**, **5**, and **50**) and inactive compounds (**44** and **45**) for the SIP₃ receptor. Model R3 Ex was applied to elucidate if it could efficiently identify the antagonists among them.

As a result, model R3 Ex identified only **5** as an active compound among the 18 derivatives in Figure 10. Although **4** and **50** were reported to be SIP₃ receptor antagonists, their K_i values were less than one-tenth that of **5**. Thus, the results based on the pharmacophore model are consistent with the reported SIP₃ receptor antagonist actives. The terminal alkyl group of the C7 alkyl chain in **5** was superimposed on *hydrophobic* 3.11, as in **8** (Figure 11a). Meanwhile, **44**, with a C6 alkyl chain, which was reported to not act as an SIP₃ receptor agonist or antagonist, had an alkyl chain that was too short to reach *hydrophobic* 3.11 (Figure 11b) and thus showed a lower fit value than **5**. In addition, since **4** with a C8 alkyl chain, which is one carbon longer than that in **5**, also shows a lower fit value than **5**, it may have reduced the potency due to repulsion between excluded volumes.

The potency for SIP₃ receptor antagonism was reported to differ significantly among **44**, **5**, and **4** with C6, C7, and C8

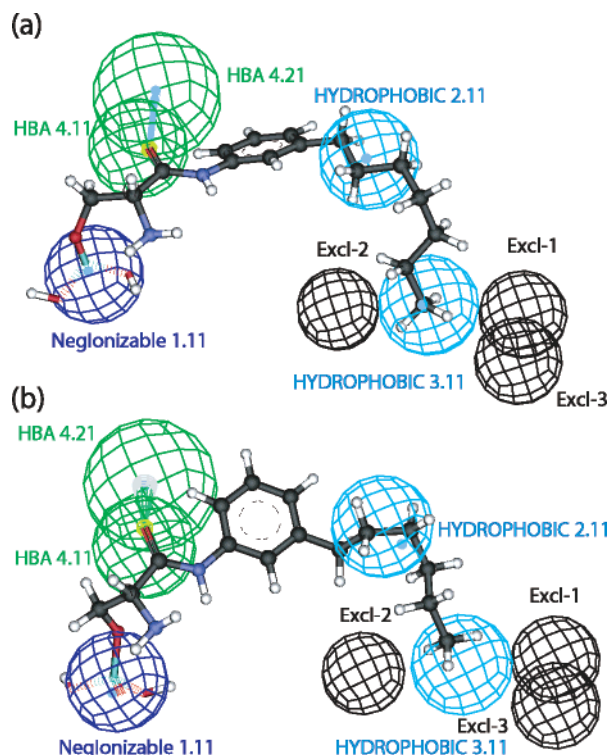


Figure 11. Overlap between model R3 Ex and **5** with a C7 alkyl chain and **44** with a C6 alkyl chain. (a) **5** with model R3 Ex. (b) **44** with model R3 Ex. Since **5** with model R3 Ex shows a better fit than **44**, the terminal alkyl group of the C7 alkyl chain in **5** is assumed to be important for SIP₃ receptor antagonism. This perspective is consistent with their reported K_i values.

alkyl chains, respectively.³² As mentioned, the pharmacophore model provided one possible explanation for the sensitivity of these antagonist activities for the SIP₃ receptor. This perspective is consistent with our results. In addition, the antagonist activities of these compounds for the SIP₁ receptor were reported to not be influenced by the length of their alkyl chains.³² If we consider all of these results, while the pharmacophore model has not clarified the role of *hydrogen-bond acceptor* feature, at least the hydrophobic space including *hydrophobic* 2.11 and 3.11 features surrounded by excluded volumes is assumed to play the key important role for SIP₃ receptor antagonism.

Conclusion

Our results demonstrate that a pharmacophore model constructed by common feature-based alignment can be used for scaffold-hopping to efficiently identify diverse active compounds prior to biological testing. For that purpose, the GH score based on well-selected structures was useful as a metric for determining relevant excluded volumes and obtaining a reliable pharmacophore model. Isophthalic acid derivatives were identified as novel and potent SIP₃ receptor antagonists by a pharmacophore-based database search. On the basis of a consideration of structure–activity relationships, the 3,4-dialkoxybenzophenone scaffold was found to be a potent component of an SIP₃ receptor antagonist. The key feature for SIP₃ receptor antagonism which was described by *hydrophobic* features with excluded volumes was assumed to be different from the space occupied by the alkyl chain of SIP. Further studies are underway to identify selective SIP₃ receptor antagonists by pharmacophore-based design.

Experimental Section

Pharmacophore Modeling Studies. Pharmacophore models were generated with the HipHop module of Catalyst 4.9^{38,53} on an Intel-based PC running the Red Hat Enterprise Linux WS 3.0 operating system. Conformational models of all compounds were calculated using an energy cutoff of 15 kcal in Best mode. The number of conformers generated for each molecule was limited to a maximum of 255. Pharmacophore generation was conducted by the feature-based alignment of five active compounds: **6**, **8**, **10**, **11**, and **12**. Compound **8** was used as the central compound to generate pharmacophore models. The database used for GH score-based validation was constructed using Catalyst/catDB, and the maximum number of conformers per compound was set at 255 in Best mode.

Conformers of (2*R*)- and (2*S*)-**13** were generated with the above-mentioned conditions and 182 and 204 conformers were obtained, respectively. All conformations in (2*R*)- or (2*S*)-**13** were aligned with the pharmacophore model derived from the training set consisting of (2*R*)- or (2*S*)-derivatives, respectively. All superimposed 3D-coordinates of C13 of **13** in the pharmacophore models were exported using the Catalyst/citest command and converted into the excluded volumes (spheres) with radius 1.5 Å. The seven active compounds of the training set were also aligned with the pharmacophore models, and the 3D-coordinates of all atoms were exported using the Catalyst/citest command and converted into atom spheres with radius 1.5 Å. Excluded volumes that overlapped any atom sphere with radius 1.5 Å were omitted because of inappropriate 3D-coordinates for an excluded volume. Appropriate excluded volumes that improved the GH score were identified and incorporated into the pharmacophore model. The Minimum Fit for Search option of the pharmacophore model was set at 3.6, which was 90% of the maximum fit value. A database search was conducted with a fast flexible search.

Stable Expression of S1P Receptors in CHO-K1 Cells. CHO-K1 cells (Dainippon Sumitomo Pharma) were transfected using Lipofectamine 2000 (Invitrogen) with pcDNA3.1 (+) plasmid (Sigma-Aldrich) encoding the human S1P₁ or S1P₃ receptor and neomycin (G418) resistance or pFLAG-CMV9 plasmid (Sigma-Aldrich) encoding the human S1P₂ receptor and G418 resistance. Cells were grown in Ham's F12K medium supplemented with G418 (200 µg/mL, Sigma-Aldrich).

Measurement of Intracellular Calcium Mobilization. A FlexStation II fluorimeter (Molecular Devices) was used to measure the intracellular calcium concentration ([Ca²⁺]_i) in CHO-K1 cells that had been stably transfected with S1P₁, S1P₂, or S1P₃ receptors. Suspensions of cells in a nutrient mixture of Ham's medium containing 1% (v/v) fetal bovine serum (FBS) were loaded (20 000 cells/well) in 96-well, clear-bottom black microplates (Greiner Bio-One) and left for 24 h at 37 °C. The cells were washed and then incubated in Ham's medium for 24 h at 37 °C.

Cells were washed and dye-loaded with Fura2-AM ester (5 µM, Molecular Probes) in Hanks balanced salt solution (HBSS) containing sulfinpyrazone (0.25 mmol/L, Sigma-Aldrich) for 60 min at 37 °C. After cell monolayers were washed with HBSS containing sulfinpyrazone (0.25 mmol/L), loading buffer was added, and cells were exposed to the test compound for 5 min. After S1P was added (10 nmol, Sigma-Aldrich), fluorescence (excitation at 335 and 362 nm; emission at 505 and 512 nm) was immediately measured for 45 s at intervals of 3 s with a FlexStation II. A curve-fitting algorithm (Model 08: Sigmoidal Inhibition Curve) implemented in MDL Assay Explorer⁵⁴ was used to calculate the IC₅₀ value.

Chemistry. General Information. Melting points were determined with a Yanako MP-500V micro melting point apparatus (uncorrected), and ¹H and ¹³C NMR spectra were recorded on a JEOL JNM-AL-300, using CDCl₃ or DMSO-*d*₆ as solvents, with Me₄Si as an internal standard. Mass spectra were recorded on either a JEOL HX-110A (FAB) or Finnigan LCQ (ESI). Elemental analysis (C, H, N) was performed on a Carloerba EA-1108 at Tokyo Chemical Industry Co., Ltd. Reactions were monitored by TLC analysis using E. Merck silica gel 60F₂₅₄ thin layer plates. Flash

chromatography was carried out on E. Merck Kieselgel 60 (230–400 mesh) silica gel.

5-Chlorosulfonyl Isophthalic Acid Dimethyl Ester (17). To a suspension of dimethyl 5-sulfoisophthalate sodium salt (5.70 g, 19.24 mmol) in CCl₄ (250 mL) were added PCl₅ (20.23 g, 96.21 mmol) and diisopropylamine (2.5 mL, 17.66 mmol). The mixture was refluxed for 6 h, diluted with H₂O at 0 °C, extracted with CHCl₃, washed sequentially with H₂O and brine, dried over Na₂SO₄, filtered, and concentrated. Recrystallization from diethylether/hexane gave **17** as a colorless powder (3.16 g, 56%). mp 91–92 °C; ¹H NMR (300 MHz, CDCl₃) δ 9.00 (s, 1H), 8.85 (s, 2H), 4.03 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 163.88, 145.13, 136.48, 132.66, 131.57, 53.19; MS (FAB) *m/z* 292 (M⁺); Anal. (C₁₀H₉ClO₆S) calcd, C 41.04; H 3.10; found, C 41.35, H 3.10.

5-Hydrazosulfonyl Benzene-1,3-Carboxylic Acid Dimethyl Ester (19). A solution of hydrazine hydrate (3.28 g, 65.60 mmol) in EtOH (50 mL) was added to a stirred **17** (3.84 g, 13.12 mmol) in CHCl₃ (100 mL). The mixture was stirred for 6 h, diluted with CHCl₃, washed sequentially with H₂O and brine, dried over Na₂SO₄, filtered, and concentrated. Recrystallization from diethylether/hexane gave **19** as colorless needles (2.11 g, 56%). mp 142–143 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.91 (s, 1H), 8.73 (d, *J* = 1.65 Hz, 2H), 4.00 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 164.56, 139.10, 135.04, 132.95, 132.09, 52.98; MS (FAB) *m/z* 288 (M⁺); Anal. (C₁₀H₁₂N₂O₆S) calcd, C 41.66; H 4.20; N, 9.72; found, C, 41.72; H, 4.09; N, 9.55.

3-Methoxycarbonylbenzenesufohydrazide (20). A suspension of 5-chlorosulfonylbenzoic acid (1.10 g, 5.00 mmol) in thionyl chloride (5 mL) was refluxed for 2 h. The mixture was concentrated, dissolved in methanol (10 mL), stirred for 15 min, and concentrated. The residue of **18** was dissolved in THF (10 mL). To the solution was added hydrazine hydrate (0.36 mL, 7.50 mmol). The mixture was stirred for 30 min at 0 °C, diluted with CHCl₃, washed sequentially with H₂O and brine, dried over Na₂SO₄, filtered, and concentrated. Recrystallization from chloroform/diethylether gave **20** as a colorless powder (367 mg, 32%). mp 87–89 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.55 (brs, 1H), 8.33 (s, 1H), 8.19 (d, *J* = 7.71 Hz, 1H), 8.04 (d, *J* = 7.94 Hz, 1H), 7.76 (dd, *J* = 7.94, 7.71 Hz, 1H), 4.21 (brs, 2H), 3.90 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.36, 137.19, 134.30, 132.23, 131.41, 129.57, 129.21, 52.70; MS (FAB) *m/z* 230 (M⁺); Anal. (C₈H₁₀N₂O₄S) calcd, C, 41.73; H, 4.38; N, 12.17; found, C, 42.08; H, 4.26; N, 11.95.

4-Cetyloxy-3-methoxybenzophenone (21). To a solution of 4-hydroxy-3-methoxybenzophenone⁵⁵ (342 mg, 1.50 mmol) in DMF (10 mL) were added sodium carbonate (249 mg, 1.80 mmol) and cetyl bromide (0.69 mL, 2.25 mmol). The mixture was stirred for 2 h at 80 °C, diluted with ethyl acetate, washed sequentially with H₂O and brine, dried over Na₂SO₄, filtered, and concentrated. Column chromatography (10:1 hexane:ethyl acetate) provided **21** as a colorless powder (605 mg, 89%). mp 57–58 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.75 (dd, *J* = 8.63, 1.65 Hz, 2H), 7.59–7.54 (m, 1H), 7.50–7.45 (m, 3H), 7.36 (dd, *J* = 8.42, 2.01 Hz, 1H), 6.88 (d, *J* = 8.42 Hz, 1H), 4.09 (t, *J* = 7.03 Hz, 2H), 3.93 (s, 3H), 1.89 (dt, *J* = 7.03, 6.89 Hz, 2H), 1.49–1.42 (m, 2H), 1.38–1.22 (m, 24H), 0.88 (t, *J* = 6.61 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 195.64, 152.70, 149.21, 138.37, 131.81, 129.87, 129.72, 128.15, 125.53, 112.40, 110.76, 69.09, 56.11, 31.93, 29.69, 29.66, 29.60, 29.55, 29.37, 28.96, 25.91, 22.70, 14.13; MS (FAB) *m/z* 452 (M⁺); Anal. (C₃₀H₄₄O₃) calcd, C, 79.60; H, 9.80; found, C, 79.35; H, 10.0.

4-Octyloxy-3-methoxybenzophenone (22). Compound **22** was prepared according to the procedure described for **21** but with octyl bromide instead of cetyl bromide. Compound **22** was obtained as a colorless powder (604 mg, quant). mp 34–35 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.76 (dd, *J* = 8.44, 1.47 Hz, 2H), 7.59–7.54 (m, 1H), 7.50–7.45 (m, 3H), 7.36 (dd, *J* = 8.44, 2.02 Hz, 1H), 6.88 (d, *J* = 8.44 Hz, 1H), 4.09 (t, *J* = 6.97 Hz, 2H), 3.93 (s, 3H), 1.87 (dt, *J* = 6.97, 6.79 Hz, 2H), 1.49–1.46 (m, 2H), 1.33–1.23 (m, 8H), 0.89 (t, *J* = 6.97 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 195.38, 152.61, 149.12, 138.24, 131.65, 129.76, 129.56, 128.00, 125.37, 112.32, 110.70, 68.94, 55.94, 31.66, 29.20, 29.07, 28.86,

2H), 6.68 (d, $J = 8.39$ Hz, 1H), 6.63 (dd, $J = 8.39, 1.80$ Hz, 1H), 3.97 (t, $J = 6.99$ Hz, 2H), 3.95 (s, 3H), 3.86 (s, 3H), 1.81 (dt, $J = 7.70, 6.58$ Hz, 2H), 1.46–1.44 (m, 2H), 1.40–1.21 (m, 24H), 0.88 (t, $J = 6.37$ Hz, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 165.38, 155.01, 150.65, 149.20, 146.37, 139.14, 134.07, 131.90, 131.29, 131.05, 130.16, 129.73, 129.23, 128.82, 128.27, 122.17, 111.30, 109.43, 68.94, 55.93, 52.61, 31.93, 29.69, 29.59, 29.54, 29.37, 28.99, 25.90, 22.70, 14.13; MS (ESI) m/z 664 (M^+); Anal. ($\text{C}_{38}\text{H}_{52}\text{N}_2\text{O}_6\text{S}$) calcd, C, 68.64; H, 7.88; N, 4.21; found, C, 68.73; H, 7.99; N, 4.05.

(E)-5-[1-(4-Cetyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono isophthalic Acid (30E). To a solution of **25E** (108 mg, 0.15 mmol) in THF (3 mL) and methanol (1 mL) was added 1 N NaOH (1 mL). The mixture was stirred for 19 h, diluted with CHCl_3 , washed with 1 N hydrochloric acid and brine, dried over Na_2SO_4 , filtered, and concentrated. Recrystallization from isopropyl alcohol gave **30E** as a pale yellow powder (97 mg, 93%). mp 161–163 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 13.82 (brs, 2H), 10.65 (brs, 1H), 8.65 (s, 3H), 7.43–7.30 (m, 5H), 7.06 (d, $J = 8.80$ Hz, 1H), 6.77–6.73 (m, 2H), 3.99 (t, $J = 6.54$ Hz, 2H), 3.69 (s, 3H), 1.73 (dt, $J = 7.09, 6.54$ Hz, 2H), 1.42–1.19 (m, 26H), 0.83 (t, $J = 7.09$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.42, 155.66, 149.16, 148.72, 139.78, 137.29, 133.63, 132.28, 132.14, 129.91, 128.26, 127.70, 124.18, 121.98, 112.57, 112.43, 68.11, 55.62, 31.32, 29.07, 28.77, 28.74, 25.58, 22.12, 13.98; MS (ESI) m/z 693 (MH^-); Anal. ($\text{C}_{38}\text{H}_{50}\text{N}_2\text{O}_8\text{S}$) calcd, C, 65.68; H, 7.25; N, 4.03; found, C, 65.90; H, 7.36; N, 3.98.

(Z)-5-[1-(4-Cetyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono isophthalic Acid (30Z). Compound **30Z** was prepared according to the procedure described for **30E** but with **25Z** instead of **25E**. Compound **30Z** was obtained as a colorless powder (53 mg, 85%). mp 179–181 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 13.81 (brs, 2H), 10.44 (brs, 1H), 8.65 (s, 3H), 7.51–7.49 (m, 2H), 7.23–7.19 (m, 2H), 7.10 (d, $J = 1.71$ Hz, 1H), 6.82 (d, $J = 8.32$ Hz, 1H), 6.45 (dd, $J = 8.32, 1.71$ Hz, 2H), 3.90 (t, $J = 6.51$ Hz, 2H), 3.68 (s, 3H), 1.66 (dt, $J = 6.59, 6.41$ Hz, 2H), 1.38–1.16 (m, 26H), 0.83 (t, $J = 7.09$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.43, 156.57, 150.05, 148.65, 139.74, 133.64, 132.53, 132.35, 132.23, 129.42, 129.14, 128.91, 128.56, 122.05, 111.65, 108.62, 68.07, 55.04, 31.29, 29.02, 28.95, 28.70, 28.57, 25.43, 22.09, 13.96; MS (ESI) m/z 693 (MH^-); Anal. ($\text{C}_{38}\text{H}_{50}\text{N}_2\text{O}_8\text{S}$) calcd, C, 65.68; H, 7.25; N, 4.03; found, C, 65.89; H, 7.32; N, 3.97.

(E)-5-[1-(4-Octyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono isophthalic Acid (31E). Compound **31E** was prepared according to the procedure described for **30E**, but with **26E** instead of **25E**. Compound **31E** was obtained as a pale yellow powder (231 mg, 82%). mp 190–192 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 10.65 (brs, 1H), 8.66 (s, 3H), 7.41–7.32 (m, 5H), 7.07 (d, $J = 8.81$ Hz, 1H), 6.77–6.75 (m, 2H), 4.01 (t, $J = 6.42$ Hz, 2H), 3.70 (s, 3H), 1.75 (dt, $J = 6.42, 6.06$ Hz, 2H), 1.44–1.42 (m, 2H), 1.40–1.24 (m, 8H), 0.87 (t, $J = 6.97$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.47, 155.68, 149.17, 148.71, 139.77, 137.31, 133.69, 132.33, 132.17, 129.95, 128.30, 127.74, 124.19, 122.02, 112.55, 112.40, 68.12, 55.64, 31.32, 28.75, 25.62, 22.15, 14.04; MS (ESI) m/z 581 (MH^-); Anal. ($\text{C}_{30}\text{H}_{34}\text{N}_2\text{O}_8\text{S}$) calcd, C, 61.84; H, 5.88; N, 4.81; found, C, 62.00; H, 5.88; N, 4.68.

(Z)-5-[1-(4-Octyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono isophthalic Acid (31Z). Compound **31Z** was prepared according to the procedure described for **30E** but with **26Z** instead of **25E**. Compound **31Z** was obtained as a colorless powder (421 mg, 91%). mp 209–210 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 10.44 (brs, 1H), 8.66 (s, 3H), 7.52–7.50 (m, 3H), 7.24–7.21 (m, 2H), 7.11 (d, $J = 2.02$ Hz, 1H), 6.84 (d, $J = 8.44$ Hz, 1H), 6.47 (dd, $J = 8.44, 2.02$ Hz, 1H), 3.92 (t, $J = 6.61$ Hz, 2H), 3.69 (s, 3H), 1.68 (dt, $J = 7.34, 6.61$ Hz, 2H), 1.40–1.22 (m, 10H), 0.85 (m, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.46, 156.61, 150.06, 148.67, 139.74, 133.68, 132.55, 132.37, 132.25, 129.44, 129.14, 128.94, 128.58, 122.08, 111.66, 108.61, 68.09, 55.05, 31.25, 28.71, 28.65, 28.59, 25.48, 22.09, 13.99; MS (ESI) m/z 581 (MH^-); Anal. ($\text{C}_{30}\text{H}_{34}\text{N}_2\text{O}_8\text{S}$) calcd, C, 61.84; H, 5.88; N, 4.81; found, C, 61.65; H, 5.82; N, 4.71.

(E)-5-[1-(4-Butyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono isophthalic Acid (32E). Compound **32E** was prepared according to the procedure described for **30E** but with **27E** instead of **25E**. Compound **32E** was obtained as a colorless powder (632 mg, 92%). mp 196–199 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 10.65 (brs, 1H), 8.66 (s, 3H), 7.39–7.32 (m, 5H), 7.08 (d, $J = 8.81$ Hz, 1H), 6.77–6.74 (m, 2H), 4.02 (t, $J = 6.42$ Hz, 2H), 3.71 (s, 3H), 1.73 (dt, $J = 6.79, 6.42$ Hz, 2H), 1.49–1.42 (m, 2H), 0.95 (t, $J = 7.34$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.41, 155.68, 149.16, 148.71, 139.77, 137.28, 133.62, 132.26, 132.14, 129.92, 128.27, 127.70, 124.17, 121.98, 112.55, 112.41, 67.77, 55.62, 30.76, 18.79, 13.69; MS (ESI) m/z 525 (MH^-); Anal. ($\text{C}_{26}\text{H}_{26}\text{N}_2\text{O}_8\text{S}$) calcd, C, 59.31; H, 4.98; N, 5.32; found, C, 58.99; H, 4.93; N, 5.16.

(E)-5-[1-(4-Cetyloxyphenyl)-1-phenylmethylene]sulfohydrazono isophthalic Acid (33E). Compound **33E** was prepared according to the procedure described for **30E** but with **28E** instead of **25E**. Compound **33E** was obtained as a pale yellow powder (13 mg, 26%). mp 175–176 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 13.82 (brs, 2H), 10.64 (brs, 1H), 8.66 (brs, 3H), 7.42–7.26 (m, 5H), 7.17 (d, $J = 7.71$ Hz, 2H), 7.03 (d, $J = 7.71$ Hz, 2H), 4.01 (t, $J = 6.40$ Hz, 2H), 1.73–1.69 (m, 2H), 1.46–1.14 (m, 26H), 0.84 (t, $J = 6.71$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.38, 159.52, 155.92, 139.62, 137.45, 132.20, 130.65, 128.23, 127.72, 124.16, 114.46, 67.56, 31.26, 29.01, 28.98, 28.71, 28.67, 28.58, 25.48, 22.06, 13.93; MS (ESI) m/z 663 (MH^-); Anal. ($\text{C}_{37}\text{H}_{48}\text{N}_2\text{O}_7\text{S}$) calcd, C, 66.84; H, 7.28; N, 4.21; found, C, 66.73; H, 7.50; N, 4.29.

(E)-3-[1-(4-Cetyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono benzoic Acid (34E). Compound **34E** was prepared according to the procedure described for **30E**, but with **29E** instead of **25E**. Compound **34E** was obtained as a colorless powder (54 mg, 83%). mp 134–135 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 13.50 (brs, 1H), 10.56 (brs, 1H), 8.50 (s, 1H), 8.20 (d, $J = 7.83$ Hz, 1H), 8.13 (d, $J = 7.72$ Hz, 1H), 7.77 (dd, $J = 7.83, 7.72$ Hz, 1H), 7.39–7.28 (m, 5H), 7.06 (d, $J = 7.94$ Hz, 1H), 6.74–6.72 (m, 2H), 4.00 (t, $J = 6.30$ Hz, 2H), 3.70 (s, 3H), 1.72 (dt, $J = 6.80, 6.32$ Hz, 2H), 1.45–1.19 (m, 26H), 0.84 (t, $J = 6.80$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 166.11, 154.98, 149.06, 148.76, 139.35, 137.32, 133.45, 131.64, 131.53, 129.75, 129.66, 128.64, 128.26, 127.55, 124.28, 121.85, 112.63, 112.37, 68.09, 55.60, 31.30, 29.06, 28.71, 25.57, 22.10, 13.97; MS (ESI) m/z 649 (MH^-); Anal. ($\text{C}_{37}\text{H}_{50}\text{N}_2\text{O}_6\text{S}$) calcd, C, 68.28; H, 7.74; N, 4.30; found, C, 68.23; H, 7.80; N, 4.25.

(Z)-3-[1-(4-Cetyloxy-3-methoxyphenyl)-1-phenylmethylene]-sulfohydrazono benzoic Acid (34Z). Compound **34Z** was prepared according to the procedure described for **30E**, but with **29Z** instead of **25E**. Compound **34Z** was obtained as a colorless powder (45 mg, 69%). mp 72–74 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 13.49 (brs, 1H), 10.37 (brs, 1H), 8.49 (s, 1H), 8.20 (d, $J = 7.80$ Hz, 1H), 8.14 (d, $J = 7.69$ Hz, 1H), 7.76 (dd, $J = 7.80, 7.69$ Hz, 1H), 7.51–7.49 (m, 3H), 7.22–7.19 (m, 2H), 7.04 (d, $J = 1.78$ Hz, 1H), 6.82 (d, $J = 8.37$ Hz, 1H), 6.47 (dd, $J = 8.37, 1.78$ Hz, 1H), 3.90 (t, $J = 6.42$ Hz, 2H), 3.67 (s, 3H), 1.67 (dt, $J = 6.57, 6.42$ Hz, 2H), 1.38–1.17 (m, 26H), 0.84 (t, $J = 6.82$ Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 166.14, 149.93, 148.59, 139.20, 137.09, 133.48, 132.61, 131.98, 131.48, 129.52, 129.37, 128.95, 128.81, 128.57, 122.31, 121.79, 111.71, 108.99, 68.08, 55.13, 31.29, 29.02, 28.96, 28.70, 28.57, 25.44, 22.10, 13.97; MS (ESI) m/z 649 (MH^-); Anal. ($\text{C}_{37}\text{H}_{50}\text{N}_2\text{O}_6\text{S}$) calcd, C, 68.28; H, 7.74; N, 4.30; found, C, 68.30; H, 7.80; N, 4.28.

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Supporting Information Available: One hundred seven structures in the database for assessing the GH score and the structures of 36 assayed compounds. All elemental analyses (C, H, N) for synthesized compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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